Half Yearly Examination 2022-23 Class -12th

Subject - Chemistry

Time:	- 3:00 h	กร		•	M.M.70		
Instruc	tion						
(1)	Question 1 to 4 are objective type question, carries 28 marks [ea						
			marks]				
(2)	Question No 5 to 12, each question carries 2 marks						
(3)	Questi	Question No 13 to 16, each question carries 3 marks					
(4)	Questi	estion No 17 carry 4 marks					
(5)			18 and 19 carries 5 ma		10000 00000 000 - 0		
(6)	Interna	nal choice is given in every question from ques. No 5 to 19					
Q. 1	Choos	e the c	orrect option		(7)		
	(i) ⁻	The n	umber of moles of solu	ite present in 10	000g of solvent is		
		called					
		(a)	Molarity	(b)	Molality		
		(c)	Normality	(d)	Formality		
	(ii)						
•		(a)	al				
		(b)	Increasing Nucear ch	370			
		(c)	Decreasing nuclear c				
		(d)	Decreasing screening				
	(iii) ⁽		tion state of Fe in K4		_ _ ?		
		(a)	+ 3	(b)	+2		
		(c)	+8	(d)	+ 5		
	(iv)₁		n is identified by lucas	(b)	Ether		
		(a)	Phenol	(d)	Alcohol		
	(-A)	(c)	Aldenyde				
	(v) I	A compound which gives oily nitrosoamine with nitrous acid at low temperature					
			Methyl amine	(b)	Dimethyl amine		
		(a) (c)	Trimethyl amine	(d)	Tri ethye amine		
	(vi) ^I	In human body carbohydrate is stored:					
	(41).	(a) In form of Glucose (b) In form of glycogen					
		(c)	form of fructose				
	(vii) ¹	Which	In form of starch is not found in RNA				
	()	(a)	Thymine	(b)	Uracil		
		(c)	Adenine	(d)	Guanine		

Q. 2	Fill in the Blanks				(7)		
	(i) 1 Faraday is equal to coulomb						
	(ii) Fast reactions are completed in less than seconds						
	(iii)	General electronic con	nfigur	ation of inner transition eleme	nt is		
	(iv) B.H.S. is an insecticide whose commercial name is						
	(v)	All aliphatic amines a	re mo	re than ammonia			
	(vi) Protein is a polymer of						
	(vii) cis [Pt(NH ₃) ₂ Cl ₂] complex is used as an agent						
Q. 3	. 3 Match the following						
	_	Ă		В	(7)		
	(i)	Soda water	(a)	Ethyl alcohol			
	(ii)	S.H.E.	(b)	Explosive			
		Coin metals	(c)	Solution of gas in liquid			
	(iv)	Spirit of wine	(d)				
	(v)		(e)	Zero Volt			
	(vi)	Hair fall	(f)	Cu, Ag and Au			
	(vii)	Chloropichin	(g)	Biotin			
0.4	Ancura	er in one word			(7)		
Q. 4	(i)	Write Arrhenius equa	tion ?	r .	(7)		
	(ii)	-		- ·			
	(iii)	How many unpaired electrons in Fe ³⁺ ? Name the organometallic compound which is used as an					
	(111)	artiknock compound in petrol?					
	(iv)	(iv) Which is formed on burning other?					
	'(v)						
	r(vi)	<u> </u>	sberg	's reagent?			
	(vii)	Which protein is help	_	_			
Q. 5	Define	Mole fraction ?			(2)		
Q. 3	Dellik	. Wide Haction :	OR		(-)		
	Define ozeotropes with 2 examples ?						
		-	-		(D)		
Q. 6	Write	the Faraday's first law	0.00	ectrolysis?	(2)		
	Write	ohm's law?	OR				
Q. 7					(2)		
	units: (i)	Rate = $K [No]^2$			7-1		
	(ii)	Rate = $K[C_2H_5OH]$					
	\ <i>)</i>						
		No. of the second secon					

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Q. 8	Define effective atomic number (EAN) with one example? OR	(2)
	Write formula of each of the following IUPAC names of co-ordina	ation
	compounds: (i) Tetrahydroxozincate (II)	
	(i) Tetranydroxozincate (II) (ii) Potassium tetrachlonido palladate (II)	
	(11)	
Q. 9	Define chelates with example?	(2)
	OR	
,	Give IUPAC name of the following	
	(i) $[Hg I_4]^{2}$	
	(ii) $K[Ag(CN)_2]$	
Q. 10	What is Helf- Volhard - Zelinsky reaction? OR	(2)
	Why Ketoness are less reactive than aldehyde?	
Q. 11	Ethyl amine is more basic than ammonia ? OR	(2)
	Write mustard oil reaction?	
Q. 12	Write any 2 difference between Globular and fibrous protein? OR	(2)
	What is DNA finger printing? Why it is important?	
Q. 13	4 gm urea dissolved in 100 gm of water The depression in freezing point was found to be 1.24°C calculate the molecular mass of urea OR	; ? (3)
	What is elevation in boiling point?	
	Drive an expression for it and its molecular mass deter mination?	
Q. 14	Write any 35 factor which affects rate of reaction? OR	(3)
	Prove that half life period of zero order reaction is proportional to	
	initial concentration of reactant.	/ 21
Q. 15	Write any 3 difference between d- and f= block elements?	(3)

Give reason

- (i) The Transition metals generally form coloured compounds.
- (ii) Transition metals exhibit varioable oxidation state.
- Q. 16 Distinguish between 1⁰, 2⁰ and 3⁰ alcohol by Lucas test? (3)

Give reaction:-

- (i) Reimer Tiemann reaction
- (ii) Kolbe's -Scmidt's reaction
- Q. 17 How does nucleophillic substitution reaction occure in alkye halide?

 Give its mechanism?

 (4)

OR

What is β elimination reaction? Mention Saytzeff's rule associated with.

Q. 18 What is rusting of Iron? describe electro chemical theory of rusting?(5)

OR

Derive Nernst equation for Daniel cell?

Q. 19 Give quick vinegar Method of preparation of acetic acid, Write its 2 properties and 2 uses? (5)

OR

Write down the following reactions

- (i) Rosenmund reaction
- (ii) Etaed's reaction
- (iii) Gattermann Koch synthesis
- (iv) Cannizaro Reaction
- (v) Perkin's reaction